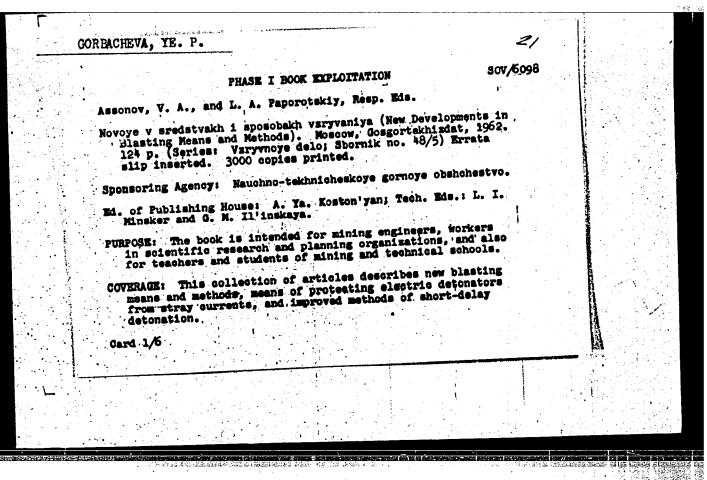
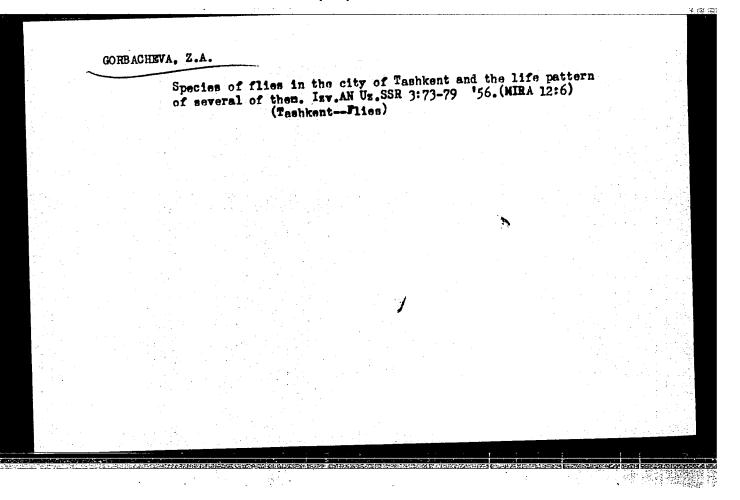
RABOVIK, Ya.I.; ORESHKINA, O.M.; GORBACHEVA, Ye.G.; KUZNETSOVA, L.A., red.

[Laboratory manual of qualitative analysis for correspondence-course students of the faculties of agronomy and zootechny] Rukovodstvo k prakticheskim zaniatiiam po kachestvennomu analizu dlia studentov-zaochnikov agronomicheskogo i zootekhnicheskogo fakul'tetov. Moskva, 1963. 170 p. (MIRA 17:8)



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Card 5/6		

Installation of an electricity sized procks. Varyv. del	1c blasting network in a no.48/5:98-101 '62.	shattering over- (MIRA 15:9)
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OCEPACHEVA, Z.A.

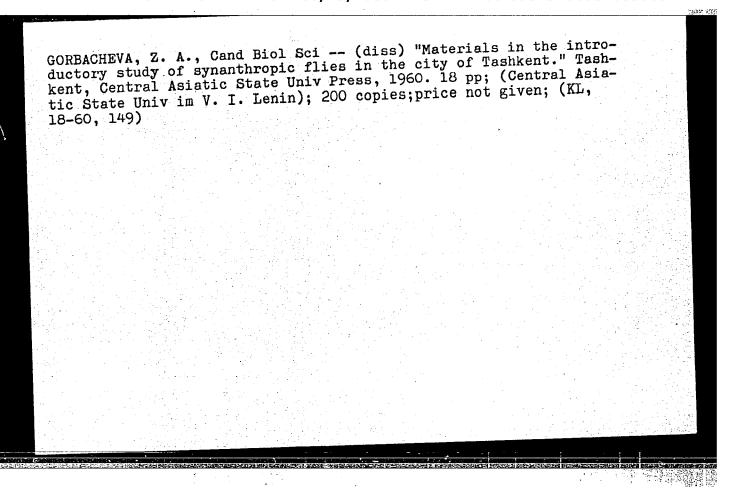
Phenology and number of generations of certain symanthropic flies in Fashkent. Med.peres. i peres.bol.supplement to no.1:75-76 '57.

Tashkent. Med.peres. i peres.bol.supplement to no.1:75-76 '10.

(MIRA 11:1)

1. Is kafedry obshchey biologii Tashkentskogo meditsinskogo institute ineni V.M.Molotova.

(TASHKIWT--FLIES)



# GORBACHEVA, Z.A.

Biology of Musca domestica vicina Macq. (housefly) in Tashkent.

Sbor.nauch.trud.TashGMI 22:45%-463 62. (MIR1 18:10)

1. Kafedra obshchey biologii Tashkentskogo gosudarstvennogo meditsinskogo instituta (zav. kafedroy - prof. M.S.Sefiyev).

USSR/Cultivated Plants. Commercial. Oil-Bearing. Sugar-Bearing.

M-5

Abs Jour

: Ref Zhur - Biol., No 7, 1958, 29870

Author

Gorbacheva, Z.I.

The Agricultural Institute of the Academy of Sciences,

Inst Uzbek SSR.

Title

Physiologico-Biochemical Changes in Cotton Leaves with

Artificial Defoliation.

Orig Pub

: V sb.: vopr. fiziol. khlopchatnika i trav. vyp. 1,

Tashkent, AN UzSSR, 1957, 95-106

Abstract

The physiologico-biochemical changes in the leaves of the cotton plant were studied when they had been treated with various chemical preparations. The content of phosphates, pigments and the concentration of cellular fluid in the leaves were determined. Aside from this, study was made of the effect of the removal of the leaf lamina on the

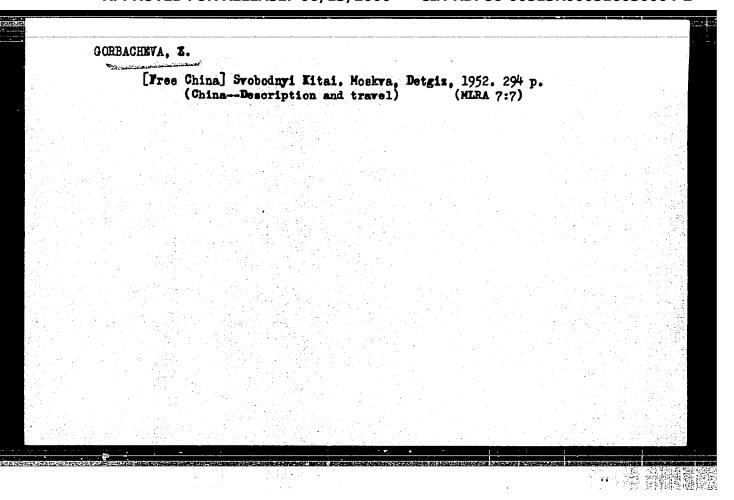
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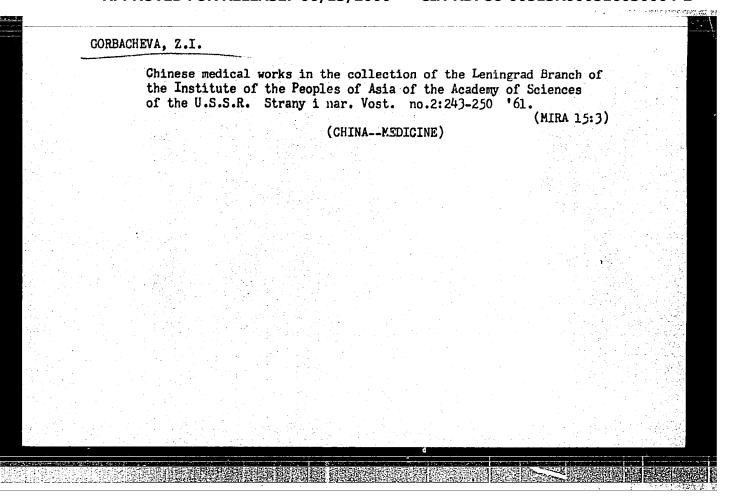
Abs Jour : Ref Zhur - Biol., No 7, 1958, 29870

> speed of the dropping off of the petioles, and various periods were tried out for pollinating cotton with calcium cyanamide. The tests were made at the experimental cotton base of the Agricultural Institute of the Academy of Sciences of the Uzbek SSR. The cotton variety 108-F and the house plant Coleus (in the experiment with the leaf lamina) were the objects under investigation. The following preparates were tried out: dusting with calcium cyanamide (40-50 kg per ha.) and spraying with the same (0.5, 2 and 5% solutions), spraying with endothal, thiocarbamide and magnesium chlorate solutions in concentrations of 0.4-0.8%. Spraying was done in the period from the end of August to October, and dusting from the middle of June to the middle of October. It was established that the removal of the leaf lamina stimulated the formation of a separating layer. The effectiveness of calcium cyanamide was reduced when applied in the later periods.

Card 2/3



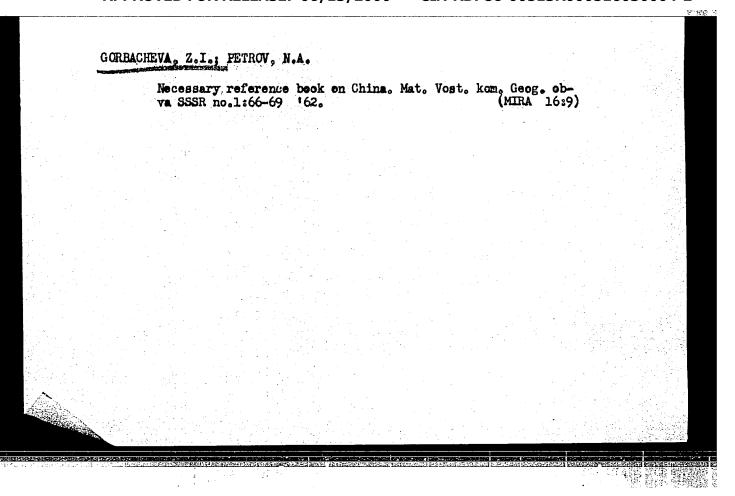
# GORBACHEVA, Z.I. First information obtained by the Chinese people about Russia. Iz ist.nauki i tekh.v stran.Vost. no.2:80-84 '61. (MIRA 14:9) (China--Relations (General) with Russia) (Russia--Relations (General) with China)



KOZLOV, Petr Kus'mich. (1863-1935); Prin. uchastiye:GORBACHEVA, Z.I.;
GUMILEV, L.N., red.; KOZLOV, V.P., red.; KOZLOVAPUSHKAREVA, Ye.V., red.; MURZAYEV, E.M., red.;
OVCHINNIKOVA, T.N., red.; SINITSIN, V.M., red.;
YUNATOV, A.A., red.; SPRIGINA, L.I., red.isd-va;
VOLKOVA, V.V., tekhn. red.

[A Russian traveller in Central Asia] Russkii puteshestvennik v TSentral'noi Asii; izbrannye trudy (k stoletiiu so dnia roshdeniia, 1863-1963). Moskva, Isd-vo AN SSSR, 1963. 522 p. (MIRA 16:10)

(Kozlov, Petr Kusimich, 1863-1935)
(Asia, Central-Discovery and exploration)



15 (6)

SOV/101-59-1-2/10

**AUTHORS:** 

Diment, P. M., Viktorenkov, V. I., Gorbachevich, I. D.,

Petrosyants, G. V., Grin'ko, A. R.

TITLE:

A Rotary Kiln with Cyclone Heat Exchangers (Vrashchayushchayasya pech! s tsiklonnymi teploobmennikami) From the Work Experience of the Spasskiy Cement Plant

(Iz opyta raboty Spasskiy tsementnogo zavoda)

PERIODICAL:

Tsement, 1959, Nr 1, pp 7 - 12 (USSR)

ABSTRACT:

The authors state that the heat of gases escaping from a rotary kiln working on a dry process is for the preparatory heating of the raw material mixture. Part of the process is carried out in the conveying calcinator, i.e. in the cyclone heat exchangers. The latter are assembled at the rear of the "Lepol" type kilns. In such kilns, prior to the calcination of clinker, the plastic raw material containing about 12% water, ought to be granulated. When using cyclone heat exchangers, the non-plastic raw materials, practically devoid of water, may also be used for calcination. The

Card 1/2

workers of Giprotsement (State Planning Institute for Cement

507/101-59-1-2/10

A Rotary Kiln with Cyclone Heat Exchangers From the Work Experience of the Spassk Cement Plant

> Industry Enterprises) and workers of the Spasskiy tsementnyy zavod (Spasskiy Cement Plant) have designed a rotary kiln provided with cyclone heat exchangers. The output of this kiln will be 14 tons per hour. An installation of cyclones working in parallel is shown in a diagram (Fig. 1). The authors state that a 3 x 60 m rotary kiln, with one cyclone line, may produce 12 - 13 tons per hour. The specific heat expenditure is about 1,000 kcal/kg of clinker. The process of calcination itself is uniform, when consistency in the feeding and quality of the raw material mixture is maintained. Stop pages in the feeding of the raw mixture layer and pronounced differences in the constitution of mixture interfere seriously with the smoothness of the process, causing a drop in efficiency. The positive results obtained with the application of cyclone heat exchangers prove the usefulness of this device. The cyclones are recommended for application in the remaining kilns of the plant in question, and as well in other plants working on the dry process. There are 2 diagrams, 1 photograph and 3 tables.

Card 2/2

GORBACHEVICH, Kirill Sergeyevich; KHABLO, Yevgeniy Petrovich; VISHNYA, L.P., red.; UNUSHKO, N.G., tekhn. red.

[Why do they have such names? On the origin of the oldest place names in Leningrad] Pochemu tak nasvany? O proiskhoshdenii starinnykh nasvanii v Leningrade. Leningrad, Lenisdat, 1962. 197 p. (MIRA 16:1)

(Leningrad-Names, Geographical)

	EVICH, Kirill Sergeyevich; FILIN, F.P., prof., otv. red.  [Russian geographical names] Russkie geograficheskie naz- vaniia. Moskva, Nauka, 1965. 63 p. (MIRA 18:8)					
	1. Chlen-	-korrespondent AN	SSSR (for Filin	<b>).</b>		

GORBACHEVSKAYA, A.S.; KASHURO, L.G.

Marble bones in very young children. Vest. rent. i rad. 40 no.6:62-63 N-D '65. (MIRA 19:1)

1. Kafedra rentgenologii Leningradskogo pediatricheskogo meditsinskogo instituta (zav. - prof. Ya.L. Shik).

S/153/60/003/004/027/040/XX B020/B054

26.1610

Barmashenko, I. B., Gorbachevskava, L. A.

TITLE:

AUTHORS:

Hydrogen Overvoltage on Porous Iron- and Iron-nickel Cathodes in Electrolysis of a Sodium Chloride Solution

PERIODICAL:

Izvestiya vysshikh uchebnykh zavedeniy. Khimiya i khimicheskaya tekhnologiya, 1960, Vol. 3, No. 4,

pp. 699 - 706

TEXT: The authors studied the process mentioned in the title at current densities of from 100 to 5000 a/m² and at temperatures of from 20 to  $80^{\circ}$ C. They measured the electrode potential by a direct method according to the usual compensation scheme with an "Etalon" ("Standard") type potentiometer and a normal Weston cell. The emf of the chain was measured in the experiments: the porous cathode investigated is a reference half-cell (oxygen-mercury electrode in 1 N NaOH solution). A NaCl solution saturated at room temperature (5.5 N or 320 g/l) was used as electrolyte. Before electrolysis, a small amount of solid salt was put into the space around the electrodes in the electrolyzer; the

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Hydrogen Overvoltage on Porous Iron- and S/153/60/003/004/027/040/XX Iron-nickel Cathodes in Electrolysis of a B020/B054 Sodium Chloride Solution

salt did not dissolve during the experiment. Fig.2 shows the curves for the dependence of hydrogen overvoltage  $(\eta_{\rm H_2})$  on the logarithm of current density (i) on porous annealed and not annealed iron- and iron-nickel electrodes at 20, 40, 60, and 80°C. Data for a smooth iron electrode are given for comparison. Fig.3 shows that the temperature coefficient of hydrogen overvoltage reaches maximum values in the temperature range from 20 to 40°C, and decreases with rising temperature. An introduction of nickel reduces  $\eta_{\rm H_2}$ , as compared with the porous iron

electrode, by 0.15 - 0.2 v, and by 0.4 v as compared with smooth iron electrodes (Table). In adaptation to conditions in industrial electrolysis, measurements were made in a vessel with flow-type electrolyte (Fig.4), in contrast to the vessel with immobile electrolyte as shown in Fig.1. Thus, it was possible to attain a constant temperature in the electrolyzer with a maximum of  $40\text{-}45^{\circ}\text{C}$  at the minimum current density investigated. Fig.5 shows the dependence of hydrogen overvoltage  $(\eta_{\text{H}_2})$  on the logarithm of current density (i) in electrolysis with a

Card 2/3

Hydrogen Overvoltage on Porous Iron- and S/153/60/003/004/027/040/XX Iron-nickel Cathodes in Electrolysis of a B020/B054 Sodium Chloride Solution

flow-type electrolyte. The curve shows that, under these conditions,  $\eta_{\rm H_2}$  on porous Fe-Ni electrodes increases with rising current density; the relation deviates from linearity at low and high current densities, as had been observed in experiments with an immobile electrolyte. The authors mention L. L. Kuz'min, L. V. Borisova, V. S. Poroykova, N. N. Voronin, D. S. Nadezhdin, V. S. Daniel'-Bek, M. A. Loshkarev, A. M. Ozerov, O. S. Ksenzhek, V. V. Stender, P. D. Lukovtsev, S. D. Levina, I. P. Fedot'yev, A. G. Pecherskaya, Zhivotinskiy, and

Stroganov. There are 5 figures, 1 table, and 23 Soviet references.

ASSOCIATION: Kiyevskiy politekhnicheskiy institut, kafedra tekhnologii

elektrokhimicheskikh proizvodstv (Kiyev Polytechnic Institute, Department for the Technology of the Electro-

chemical Industry)

SUBMITTED: October 17, 1958

Card 3/3

FILIPPOVICH, Z.S.; PETRIK, K.G., multivally about: AVEDITANOV, K.G., rukovoditel' rabot; Prinimali uchastive: KACHANOVSKAYA, Z.I.; GANTMAN, Ya.I.; KHUSID, B.S.; GORBACHEVSKAYA, M.S.

Increasing the coefficient of utilization of fresh fruit and berries in the winemaking, juice and liqueur-and-vodka industries. Trudy BNIIPPT no.4:129-144 '61. (MIRA 17:10)

SOBOLEV, V.S.; GORBACHEVSKAYA. O. N.

Aegirite in tuffite from Tertiary deposits of the Carpathian peidmont. Min. sbor. no.5:159-166 \*51. (MLRA 9:12)

1. Gosuniversitet imeni Ivana Franko i Institut geologii poleznykh iskopaemykh, Akademiya nauk USSR. (Carpathian Mountain region-Aegirite) (Carpathian Mountain region-Tuffite)

GORBACHEVSKAYA. O. N.

"Geologicopetrographical Investigation of the Liparites of the Velikiy Sholles Range in Transcarpathia." Cand Geol-Min Sci, L'vov State U, L'vov. 1954. (RZhGeol, Apr 55)

SO: Sum. No. 704, 2 Nov 55 - Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions(16).

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GORBACHEVSKIY, A. M.

Medical Instruments and Apparatus

Apparatus for blowing through fallopian tubes. Med. prom. no. 5, 1952.

Monthly List of Russian Accessions, Library of Congress, December 1952. Unclassified.

BALLYUZEK, F.V., doktor med. nauk; SKORIK, V.I., kand. med. nauk; GORBACHEVSKIY, A.M.; KVENITSKIY, G.R.

Technical equipment for regional perfusion of the extremities. Ortop. travm. i protez. 26 no.6:7-12 Je '65. (MIRA 18:8)

1. Iz khirurgicheskoy kliniki usovershenstvovaniya vrachey No.l (nachal'nik - prof. A.P. Kolesov) Voyenno-meditsinskoy ordena Lenina akademii imeni Kirova i Spetsial'nogo konstruktorskogo byuro ob"yedineniya "Krasnogvardeyets" (nachal'nik - I.Ya. Gurevich). Adres avtorov: Leningrad F-13 Fontanka, d.106, Khirurgicheskaya klinika usovershenstvovaniya vrachey No.l Voyenno-meditsinskoy akademii imeni Kirova.

	400 years of	Russian boo	k printing.	Vest.mashinostr.	44 no.3: (MIRA 17:4)	
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## GORBACHEVSKIY, F.F.

Determination of the effect of local high-frequency vibrations on the tonal state of the peripheral vessels of man [with summary in English]. Trudy ISGMI 44:208-230 158 (MIRA 11:12)

1. Kafedra gigiyeny truda a klinikoy profaabolevaniy Leningradakogo sanitarno-gigiyenicheskogo meditsinakogo instituta (sav. kafedrey - prof. Ye.TS. Andreyeva-Galanina).

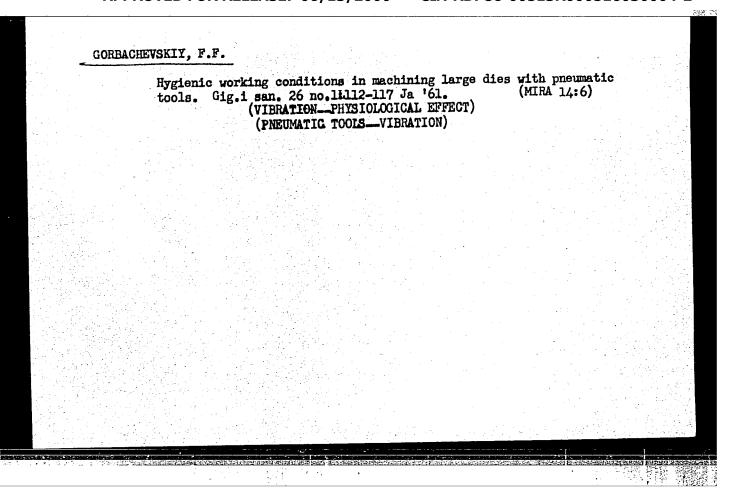
(RICOD VESSEIS, physiol.

eff. of local high frequency vibrations on tonal
state of peripheral vessles (Rus))

(VIBRATIONS, eff.

local high frequency vibrations on tonal state
of peripheral blood vessels (Rus))

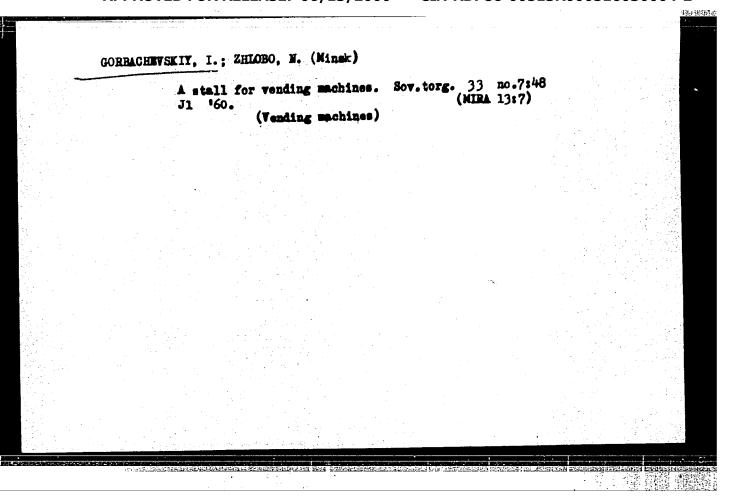
GORBACHEVSKIY, F. F., Candidate Med Sci (diss) -- "On the physiological-hygienic characteristics of high-frequency vibrations". Leningrad, 1959. 16 pp (Min Health RSFSR, Leningrad Sanitary-Hygiene Med Inst), 200 copies (KL, No 25, 1959, 140)



GORBACHEVSKIY, Fedor Fedorovich; SHAGAN, Iosif Bentsionovich; EURLOVA, L.Ya., red.; KHARASH, G.A., tekhn. red.

[Prevention of occupational diseases in the operation of gasusing boiler rooms] Preduprezhdenie professional'nykh zabolevanii pri obsluzhivanii gasifitsirovannykh kotel'nykh. Leningrad, Medgiz, 1962. 57 p. (MIRA 15:7) (Boilers-Safety measures)

# GORBACHEVSKIY, F.F. Changes in the cardiovascular system under the effect of low frequency and low intensity acoustic stimuli. Trudy LSGMI 75: 150-155 '63. (MIRA 17:4) 1. Kafedra gigiyeny truda s klinikoy professional 'nyk zabolevany (zav. kafedroy - prof. Ye.TS. Andreyeva-Galanina) Leningradskogo sanitarno-gigiyenicheskogo meditsinskogo instituta.



Waing electric vibrators for unloading freight cars. Zhel.dor.
(MIRA 12:6)

(Vibrators) (Rallroads—Freight cars)

(Loading and unloading)

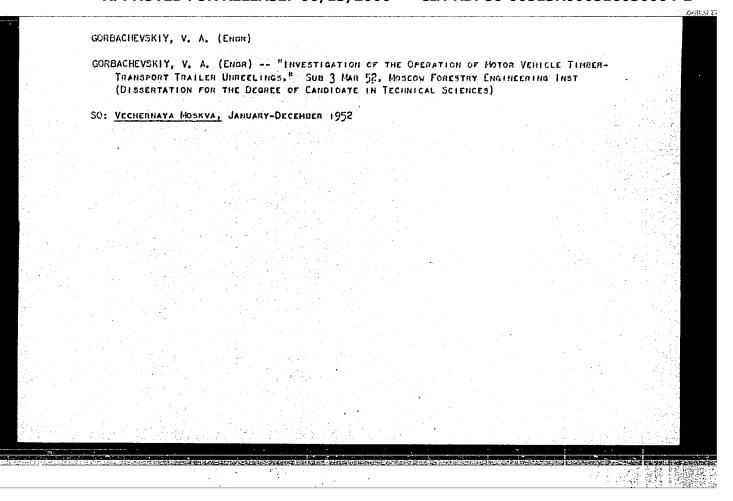
GORBACHEVSKIY, O. S.

On the Superdistance Propagation of Sound in Stretches of Deep Water

Priroda, 1951, 2 Feb, page 54

	Vehicles for 17 0 '61.	transporting long pipes (Pipe—Transportatio	. Avt. traps. (MIRA	39 no.10:15- 14:10)	
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GORBACH	IEVSKIY, V. A.	15049
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	"Automobile and Tractor Lumber-conveyer & A. Gorbachevskiy, 7 pp	Sleighs, "V.
	"Les Prom" No 8	
	Discusses trailers for single-gauge snow-covered tractor roads, for single-gauge s covered automobile roads, for trackless s tractor roads, for trackless snow-covered roads, and manufacture and use of automob tractor lumber-conveyer sleighs. Seven d grams of tractor and automobile single- a runner sleighs.	now- and ice- now-covered automobile oile and
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- 1. GORBACHEVSKIY, V.A.
- 2. USSR (600)
- 4. Steam Boilers
- 7. New steam equipment to heat engines before starting. Les.prom. 12 no.10, 1952.

9. Monthly List of Russian Accessions, Library of Congress, January 1953, Unclassified.

rab. 7 no.	11:12-14 D '53.		ailer-trucks. Mekh. trud (MERA 6:12) (EunberTransportation)	
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# GORBACHEVSKIY, V. A.

KIRYUKHIN, Anatoliy Mikhaylovich; GCBBACHEVSKIY, Viktor Andreyevich; LESHKEVICH, Andrey Ivanovich; MIKHAYLOVSKII, Turiy Vsevolodovich; GATSKEVICH, A.I., redaktor; VCROB'YEVA, N.N., redaktor; KARASIK, N.P., tekhnicheskiy redaktor

[Operation of hauling equipment] Ekspluatatsiia tiagovykh mashin.
Moskva, Goslesbumizdat, 1954. 391 p.
(Lumbering--Equipment and supplies)

YESIPCHUK, P.P.; GORBACHEVSKIY, V.A.; BALOBANOV, A.S., red.; OSOKINA, A.M., red.; izd-va; KARASIK, N.P., tekhn. red.; VOIKHOVER, R.S., tekhn. red.

[L-47 single drum winch for the S-80 tractor; "Forestry and Immber" pavilion] Odnobarabannaia lebedka L-47 dlia traktora S-80; Pavil!-on lesnaia promyshlennest! lesnoe khoziaistvo. [Moskva] M-vo lesnoi promyshl. SSSR [1956] 6 p. (MIRA 11:10)

1. Moscow. Vsesoyusnaya promyshlennaya vystavka. (Winches)

GORBACHEVSKIT, V.A.: UVAROV, N.V.: SHCHETININ, I.P., red.; MERZHANOVA,

O.W., red. isd-va; KARASIK, N.P., tekhn, red.; VOIKHOVER, P.S.,
tekhn, red.

[NAE-501 leg truck] Lesovesayi avtomobil' NAE-501. Noskva, N-vo
lesmid promyshl. SSSR, 1956. 9 p. (NIRA 11:10)

(Motortrucks)

(Notortrucks)

SOV/124-57-5-5244

Translation from: Referativnyy zhurnal. Mekhanika, 1957, Nr 5, p 20 (USSR)

AUTHOR: Gorbachevskiy, V. A.

TITLE: The Kinematic Motion of an Automobile Equipped With a Pole Trailer

in Plane Curves (Kinematika dvizheniya avtomobilya s rospuskom po

krivym v plane)

PERIODICAL: Tr. Tsentr. n.-i. in-ta mekhaniz. i energ. les. prom-sti, 1956,

Vol 3, pp 3-29

ABSTRACT: The paper analyzes the kinematic problem of the accuracy of the

tracking of automobile trailers (with straight and cross coupling) behind the tractor vehicle on a curve with a load of a considerable

length.

K. S. Kolesnikov

Card 1/1

Developing truck-hauling of logs. Mekh.trud.rab.ll no.6:20-25 Jo 157.
(MIRA 10:11)
(Lumber--Transportation)

GORRACHEVSKIY, V., kandidat tekhnicheskikh nauk.

Preliminary packet loading of lengthy freight items. Avt. transp.35 no.1:10-13 Ja '57. (MIRA 10:3)

1. TSentral'nyy nauchno-issledovatel'skiy institut mekhanisateli i energetiki. (Loading and unloading'

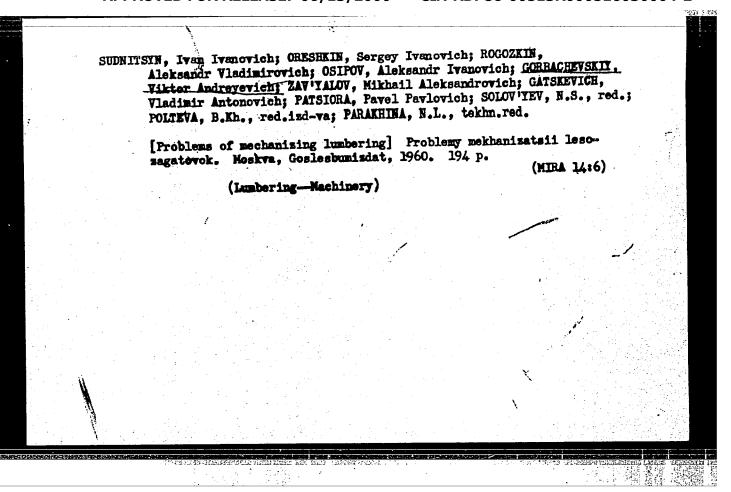
GORBACHEVSKII J. kand.tekhn.nauk; METALLIKOV, S., kend.tekhn.nauk

Using the MAZ trucks to transport timber on icy roads. Avt.transp.
(MIRA 10:12)

(Motortrucks--Gold weather operation)

(Motortrucks--Gold weather operation)

		Operating	properties	of the	MAZ-501	motortru	ck. Avt.tra	nsp. 35	ili Pr	
		no.4:24-26	Ap 157.		4		,	(MIRA	10:5)	
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GORBACHEVSKIY, Viktor Andreyevich; LESHKEVICH, Andrey Ivanovich;
MIKHAYLOVSKIY, Yuriy Vsevolodovich; SHESTAKOV, Boris
Aleksandrovich; MEDNIKOV, I.N., retsenzent; MOROZOV, K.P.,
retsenzent; KHASMAN, P.Ya., otv. red.; PLESKO, Ye.P., red.;
GRECHISHCHEVA, Z.I., tekhm. red.

[Fundamentals of lumbering and the operation of machines and mechanisms] Osnovy lesozagotovok i ekspluatatsiia mashin i mekhanizmov. V.A.Gorbachevskii i dr. Moskva, Goslcsbumizdat, 1961. 319 p. (MIRA 15:2)

GORRACHEVSKIY, Viktor Andreyevich; GAL'FERIN, Zinoviy Samoylovich
Gal'perin; KLYCHKOV, Pavel Dmitriyevich; LAKH, Yevgeniy
Ivanovich; LEKSAU, Igor' Nikolayevich; PRASOLOV, Boris
Aleksandrovich; NYZHKOV, Aleksey Nikolayevich; SUKHARNIKOV,
Ioaip Osipovich; SHESTAKOV, Boris Aleksandrovich; ALPATSKIY,
I.V., red.; PLESKO, Ye.P., red.izd-va; GRECHISHCHEVA, V.I.,
tekhn. red.

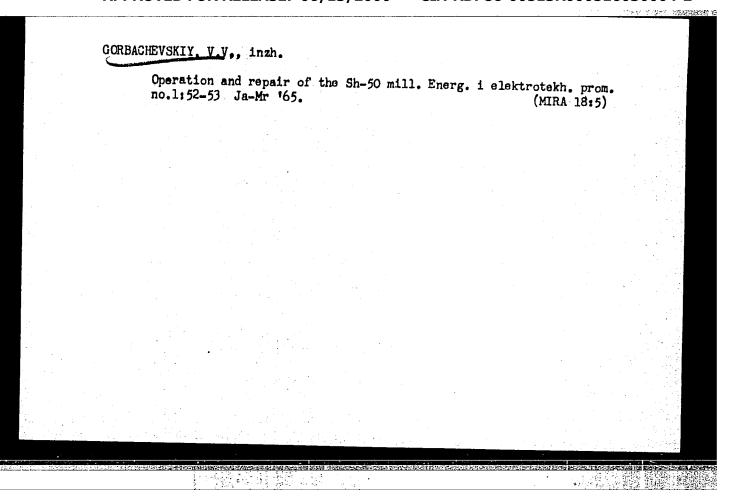
[Utilisation of logging truck transportation] Ekspluatatsiia lesovosnogo avtomobil'nogo transporta. [By] V.A.
Gorbachevskii i dr. Moskva, Goslesbumizdat, 1962. 296 p.
(MIRA 16:5)

(Lumber--Transportation) (Tractor trains)

CAL'PERIN, Z.S.; KLYCHKOV, P.D.; LAKH, Ye.I.; CORBACHEVSKIV, V.A.;
DARAGAN, L.D.; RYZHKOV, A.N.; SUKHARNIKOV, I.O.; TURASS,
A.L.; GATSKEVICH, V.A., red.

[Manual on automotive transportation of lumber] Sprayochnik po lesovoznomu avtomobil'nomu transportu. Moskva, Lesnaia promyshlennost', 1965. 446 p. (MIRA 19:1)

1. Khimki. TSentral'nyy nauchno-issledovatel'skiy institut mekhanizatsii i energetiki lesnoy promyshlennosti.



TSEYTLIN, M.A., inzh.; GORBACHEVSKIY, V.V., inzh.

Installation of pins and gunite lining in furnace screens of boilers with liquid slag removal. Energ. i elektrotekh. prom. no.3:57-58 J1-S \*65. (MIRA 18:9)

Gorbach Erskiy, YE, V.

5(2);25(1)

PHASE I BOOK EXPLOITATION

SOV/2313

Akademiya nauk SSSR. Institut mashinovedeniya

- Povysheniye stoykosti detaley mashin /sul'fidirovaniye/; sbornik statey (Increasing the Wear Resistance of Machine Parts /Sulfurization/; Collection of Articles) Moscow, Mashgiz, 1959. 126 p. Errata slip inserted. 4,500 copies printed.
- Ed. (Title page): M. M. Khrushchov, Doctor of Technical Sciences; Ed. (Inside book): A.G. Nikitin, Engineer; Tech. Ed.: V.D. El'kind; Managing Ed. for Literature on General Technical and Transport Machine Building (Mashgiz): K.A. Ponomareva, Engineer.
- PURPOSE: This collection of articles is intended for engineering and technical workers of machine-building and overhauling plants.
  - COVERAGE: This book presents results of investigations of methods to increase the resistance of machine parts to seizure. A new method of sulfurization which improves the friction behavior of cast iron and steel and an analysis of the effect of sulfurization on the antifriction properties and wear of metal are given.

Card 1/6

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These articles are the transactions of a seminar held at the Institute of Mechanical Engineering of the Academy of Sciences, USSR, in December 1956.

### TABLE OF CONTENTS:

D'yachenko, P. Ye., Doctor of Technical Sciences. Use of Sulfurization in Czechoslovakia

The author reviews the development and introduction of sulfurization in several Czech plants. The process and its advantages are described.

Vinogradov, Yu. M., Candidate of Technical Sciences. Properties of Metals Following Thermochemical Sulfurization.

The author describes investigations of sulfurization and other similar treatment carried out at the NIIKhIMMASh (Scientific Research Institute of Chemical Machinery) and gives formulas for the bath used, methods of operation, and results obtained.

Card 2/6

SOV/2313

Vaynshteyn, V.E., and Yu. M. Vinogradov, Candidates of Technical Sciences. Investigating Wear of Sulfurized Metal Surfaces by Means of Radioactive Isotopes

The authors describe an investigation carried out by the NIIKhIMMASh (Scientific Research Institute of Chemical Machinery), in which isotope S35 was used to determine the distribution of sulfur in the metal.

Somin, B.Kh., Candidate of Technical Sciences, and Ye. V. Gorbach-evskiy, Engineer, Sulfocyanation as a Means of Increasing Resistance to Seizure.

The authors describe the combined process of sulfurization and cyanation of surfaces. The mechanism and the role of both of these processes in the combined process is given.

Dombrovskaya, N.S., Doctor of Chemical Sciences, Ye. A. Alekseyeva, and N.V. Khakhlova, Engineers. Selecting Salt Baths for Sulfurization of Iron Alloys

The authors recommend the use of a salt bath as the most controllable and uniform method of sulfurization. They develop the compositions of these baths and the optimum Card 3/6

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temperatures of operation.

Zinovich, N.S., Engineer. Investigation of the Sulfurization Process

The author discusses sulfurization in the liquid bath, baths operating at medium and low temperatures, control of the process, x-ray and metallographic investigations, hardness, work-in, and wear resistance tests.

Zelenova, V.D., Engineer. X-ray Analysis of the Surface Layer of Sulfurized Specimens

The author investigated various bath compositions by x-ray analysis in order to evaluate the character of sulfurization in respect to simultaneous formation of nitrades.

Gil'man, T.P., Engineer. Sulfurization of Iron Carbide With Gas99 in mineral oil and ammonia are introduced together into the furnace. This process is a combined sulfurizing and Card 4/6

SOV/2313

105

with other sulfurization methods according to the author.

Gil'man, T.P., Engineer. Sulfurization of Bushings Made of Iron Powder by Introducing Sulphur Into the Charge
The author describes the results of experiments using a method, claimed by the author to be new. The work was carried out at Stalingrad Tractor Plant in collaboration with NATI (Automobile and Tractor Scientific Research Institute). The author stresses the advantages of this process which gives a uniform distribution of sulfides in the metal.

Smovt, M.S., Engineer. Results of Work on the Technology of the Sulfurization Process in Rostsel'mash /Rostov-na-Donu Agricultural Machinery Plant/
The author describes an investigation carried out at the Rostov plant aimed at improving wear resistance of cutting tools by sulfurization.

Lifshits, Ya. G., Candidate of Technical Sciences. Uses of Card 5/6

SOV/2313

Sulfurization in Manufacturing Agricultural Machinery In this article the author presents the results of laboratory and bench tests of sulfurized and nonsulfurized machine parts carried out by RISKhM (Rostov Institute for Agricultural Machinery) and ROSTSEL MASh.

Blokhin, M.A., P.S. Nesterenko, and A.T. Shuvayev. X-ray and Spectrum Analysis of Sulfurized Samples

The author describes an investigation of depth distribution of sulfur in type 45 steel and gray cast iron sulfurized at

Lesnykh, D.S., Candidate of Chemical Sciences. Electrosulfurization

The author presents the results obtained from sulfurizing parts in various molten salts at 240 to 270°C and in aqueous solution of salts and 50 to 75°C using electrolytic

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Card 6/6

10-20-59

ACCESSION NR: AR4018314

B/0137/64/000/001/C036/C036

SOURCE: RZh. Metallurgiya, Abs. 1G251

AUTHOR: Somin, B. Kh.; Gorbachevskiy, Ye. V.; Latsh, V. V.; Minayev, N. G.

TITLE: The influence of nickel on the sinterability of pressed powders of tungsten and molybdenum

CITED SOURCE: Tr. Kuyby\*shevsk. aviats. in-t, vy\*p. 16, 1963, 141-148

TOPIC TAGS: powder metallurgy, nickel, tungsten, molybdenum, material strength, heat-treatment

TRANSIATION: Research was conducted on the influence of Ni on sintering in an atmosphere of H<sub>2</sub> and in vacuum Mo and W in a range of 1,100-2,000 degrees for Mo and 1,100-2,500 for W, with a nickel content of 0.01-10% by weight. Density (P), microstructure, microhardness, and the parameters of the crystal network of the first phase were studied. An increase in the density of the sintered Mo with an inclusion of 0.5-1% Ni takes place as low as 1,100 degrees. At 1,300 degrees, the porosity of the samples with the above nickel content amounts to 10%. At 1,500 degrees, the effectiveness of the influence of small inclusions of Nickel on the sinterability

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ACCESSION NR: AR4018314

of Mo decreases considerably. The curves of function P of sintered W with the nickel content has a maximum equal to a 0.25% nickel content. After sintering at 1,500 degrees, the W with an admixture of 0.25-0.5% nickel amounts to 5-7%. The inclusion of nickel also leads to an increase in the microhardness of W from 250 to 600 kg/sq cm, and the microhardness of Mo from 150 to 500 kg/sq cm. The liquito 600 kg/sq cm, and the microhardness of Mo from 150 to 500 kg/sq cm. The liquito flability of nickel at 1,500 degrees is 0.3 atmospheric % in W and 1 atmospheric % fiability of nickel at 1,500 degrees is 0.3 atmospheric % in W and 1,495 degrees for in Mo. At sintering temperatures of 1,350 degrees for Mo, and 1,495 degrees for W, and a nickel content greater than 0.5% for Mo and 0.25% for W, an oozing out of W, and a nickel content greater than 0.5% for Mo and 0.25% for W, an oozing out of W, and a nickel content greater than 0.5% for Mo and 0.25% for W, an oozing out of W, and a nickel content greater than 0.5% for Mo and 0.25% for W, an oozing out of W, and a nickel content greater than 0.5% for Mo and 0.25% for W, an oozing out of W, and a nickel content greater than 0.5% for Mo and 0.25% for W, an oozing out of W, and a nickel content greater than 0.5% for Mo and 0.25% for W, an oozing out of W, and a nickel content greater than 0.5% for Mo and 0.25% for W, an oozing out of W, and a nickel content greater than 0.5% for Mo and 0.25% for W, an oozing out of W, and a nickel content greater than 0.5% for Mo and 0.25% for W, an oozing out of W, and a nickel content greater than 0.5% for Mo and 0.25% for W, an oozing out of W, and a nickel content greater than 0.5% for Mo and 0.25% for W, an oozing out of W, and a nickel content greater than 0.5% for Mo and 0.25% for W, and 0.25% for W, and 0.25% for M, and 0.25% for W, and 0.25% for M, and 0.25% for M

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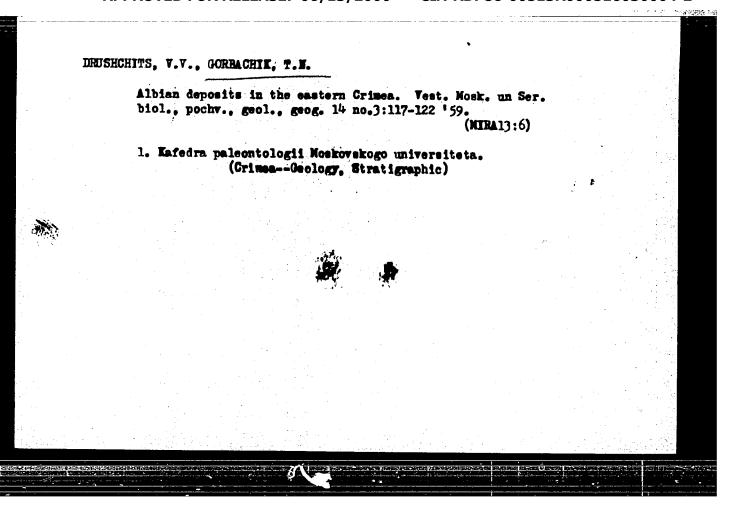
Card 2/2

GORBACHIK, T. N.

"Foraminifers of the Lower Cretaceous Deposits of the Central Ciscaucasis and the Southwestern Crimea." Moscow Order of Lenin State U imeni M. V. Lomonosov, Moscow, 1955. (Dissertation for the Degree of Candidate of Geological and Mineralogical Sciences)

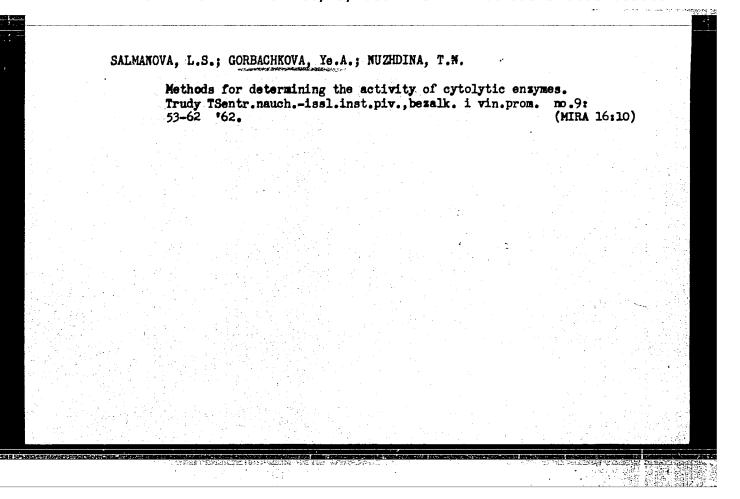
50: Knizhnava Letopis', No. 22, 1955, pp 93-105

# GORRACHIK, T.N. New foraminifer species from lower Cretaceous deposits of the Grimea and northwestern Caucasus. Falcont. zhur. no.1:78-83 (59. (MIRA 13:1) 1. Moskovskiy gosudarstvennyy universitet im. M.V. Lomonosova. (Crimea--Foraminifera, Fossil) (Pshekhi Valley--Foraminifera, Fossil)



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Biul. MOIP Otd.	rly Cretaceous geol. 37 no.6:	foraminifers	in the	Crimea.	
			(MIRA	16:8)	
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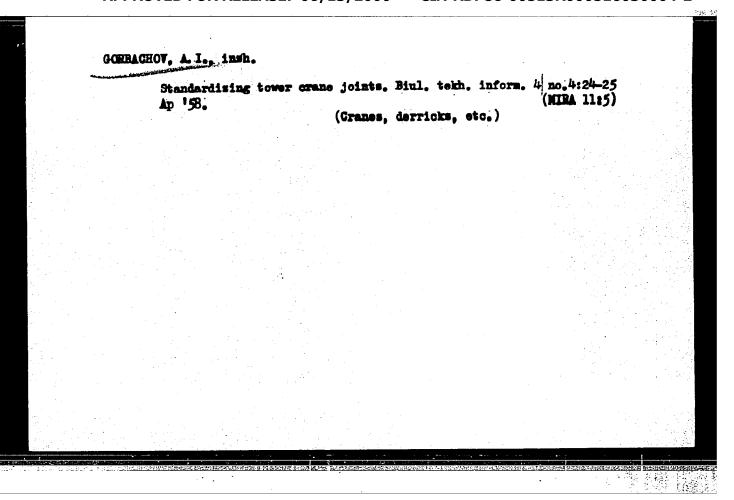
Variation and microstructure of the mural in the shell of Globigerinelloides algeriana. Paleont. zhur. no.4:32-37 '64.
1. Moskovskiy gosudarstvennyy universitet. (MIRA 18:3)

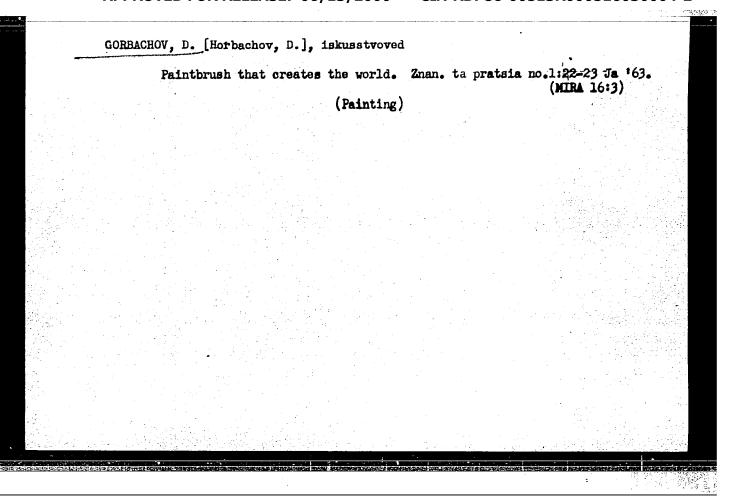


SALMANOVA, L.S.; GORBACHKOVA, Ye.A.

Carbohydrate composition of hydrolysates obtained as the result of the action of cytolytic fungi cultures on various substrates.

Trudy 2006 162. (MIRA 16:10)





Physical Principles of Instruments. Defense	of Devices and Operate Publ. House (1953)	tion of Aircraft p. 427	

USSR/General Biology. Individual Development. Embryonic Development.

Abs Jour : Ref Zhur-Biol., No 16, 1958, 71597

Author : Gorbachova, A. P., Popekhina, P. S.

: Age Determined Changes in Amino Acid Contents Title in Pig Embryos.

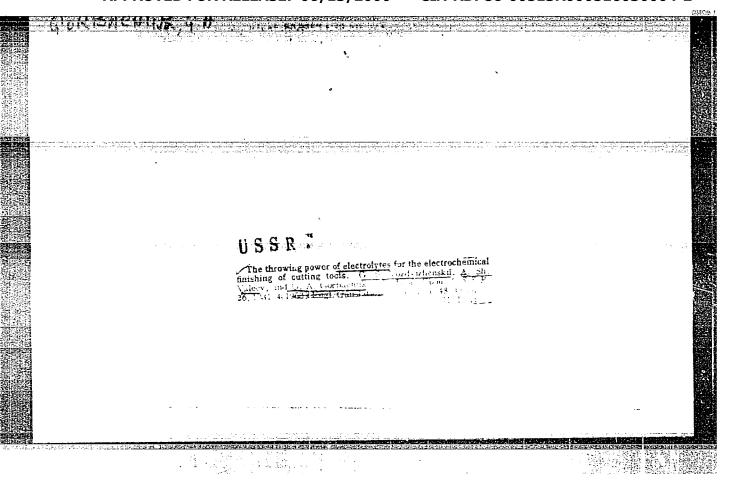
Orig Pub: Ukr. biokhin. zh., 1957, 29, No 1, 96-100

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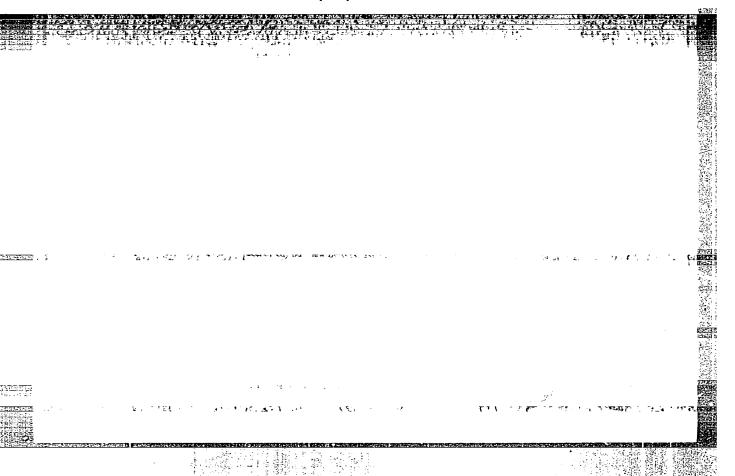
Abstract: The content of water decreases and the quantity of nineral substances increases in proportion to the growth of pig enbryos. The content of proteins in the first 40 days of development increases somewhat and then decreases sharply. Throughout embryogenesis, the ratio of different anino acids also changes. The

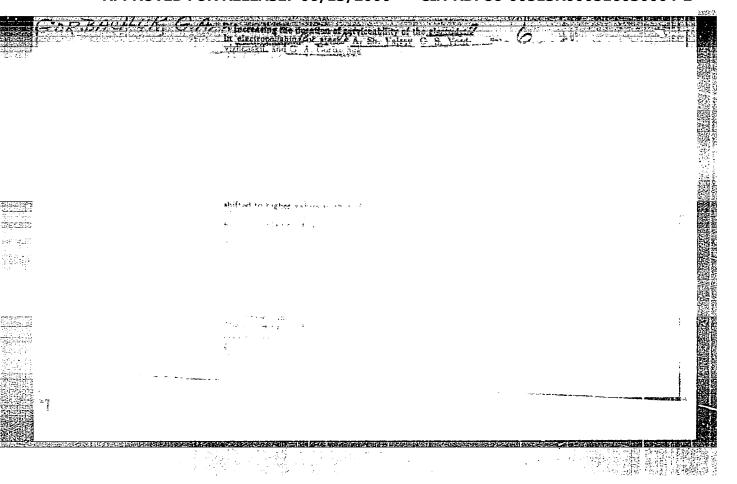
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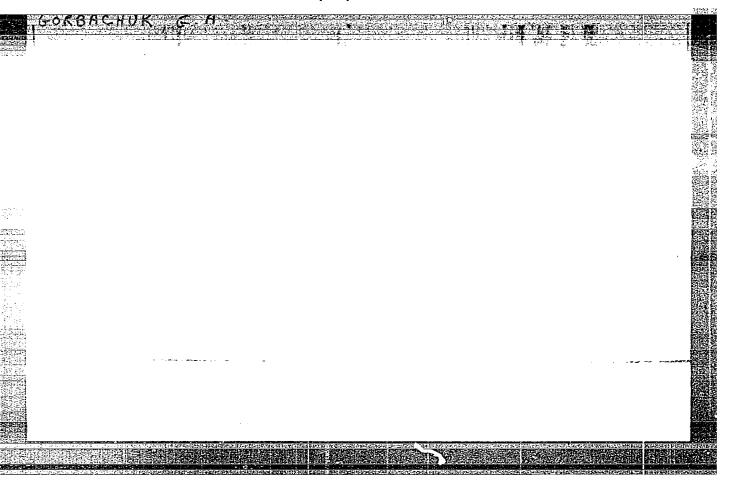
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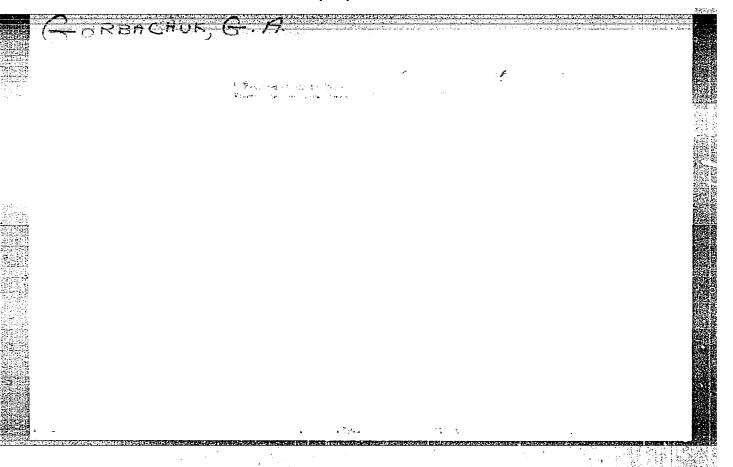


VOSDVIZHENSKIY, G.S.; VALEYEV, A.Sh.; GORBACHUK, G.A.									
	Dispersibility of electrolytes during the electrochemical processing of cutting tools. Zhur.prikl.khim. 26 no.10:1094-1096 0 '53. (MLRA 6:10) (Electrolytes) (Cutting machines) (MetalsFinishing)								
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GORBACHUK, G. A.

137-58-5-10249

Translation from: Referativnyy zhurnal, Metallurgiya, 1957, Nr 5, p 194 (USSR)

Vozdvizhenskiy, G.S., Valeyev, A.Sh., Gorbachuk, G.A. AUTHORS:

On the Mechanism of the Dissolution of Steel Upon Anodic Polar-TITLE:

ization by Low-density Currents (K voprosu o mekhanizme rastvoreniya stali pri anodnoy polyarizatsii tokami maloy plot-

PERIODICAL: Izv. Kazansk. fil. AN SSSR. Ser. khim. n., 1957, Nr 3,

pp 63-67

ABSTRACT: Results are presented of a study of the mechanism of the dissolution of steel upon anodic polarization by low-density currents in order to clarify the phenomenon of destruction of the specimen in depth without visible destruction of its surface caused by an electropolishing bath. The current efficiency (Beff) was determined by the weight loss due to anodic dissolu-

tion and spontaneous dissolution. An increase in the density of the polarizing current, all other conditions being equal, should increase the concentration of Fersalts in the anode area of the bath and reduce the concentration of oxidizer. When the temper-

Card 1/2 ature is reduced from 80 to 60°C, the rate of diffusion of the

137-58-5-10249

On the Mechanism of the (cont.)

oxidizer declines and attains its steady-state value at the very start of the process. When the densities of the polarizing current at the onset of the process are very low,  $B_{eff}$  is <100%, as it is at higher densities and under conditions of long-continued polarization. This indicates the presence not only of anodic dissolution but of some other process at the anode, possibly oxidation of trivalent  $Cr^{3+}$  ions to  $Cr^{6+}$ . The resultant data confirm the author's earlier concepts on the mechanism of breakdown of metals in electropolishing electrolytes.

Ya.L.

1. Steel--Disintergration 2. Anodes--Polarization

Card 2/2

GORBACHUK, G.A.

137-58-5-10251

Translation from: Referativnyy zhurnal, Metallurgiya, 1958, Nr 5, p 194 (USSR)

AUTHORS: Valeyev, A.Sh., Gorbachuk, G.A.

TITLE: Chemical Processes Accompanying the Dissolution of Steel in

Anodic Polarization by Low-density Currents (Khimicheskiye protsessy, soprovozhdayushchiye rastvoreniye stali pri anodnoy

polyarizatsii tokami maloy plotnosti)

PERIODICAL: Izv. Kazansk. fil. AN SSSR. Ser. khim. n., 1957, Nr 3,

pp 69-74

ABSTRACT: The results of an analytical study of the causes of the dissolution of Fe on anodic polarization by low-density currents

in electropolishing bath containing Cr acid are presented. The process of dissolution proceeds with reverse precipitation of highly disperse Fe as a loose friable mass. The processes of dissolution and reverse liberation proceed in the depth of the metal, virtually without affecting its surface layer. A study was made of the applicability of the equation Cr<sub>2</sub>O<sub>7</sub>+6Fe<sup>2+</sup>+14H+ +Fe<sup>3+</sup>+2O<sub>7</sub>+7H<sub>2</sub>O for the calculation of the consumption of

the Cr<sub>2</sub>O<sub>7</sub> inhibitor. It is shown that an anomaly in the redox process of conversion of the Cr compounds was observed. The

Card 1/2

Chemical Processes Accompanying (cont.)

usual process of reduction  $Cr^{6+} \rightarrow Cr^{3+}$  is not confirmed by the analytical data. This anomaly obviously pertains to the process of reduction of Cr on cathode segments of the microcells of the polarizing electrode.

Ya. L.

1. Steel--Disintergration 2. Anodes--Polarization

Card 2/2

SOV / 20-120-1-26/63 AUTHORS: Vozdvizhenskiy, G. S., Gorbachuk, G. A., Dezider'yev, G. P.

TITLE: On the Problem of the Mechanism of the Electrolytic Polishing

of Metals and of the Structure of the Polished Surface (K voprosu o mekhanizme elektroliticheskoy polirovki metallov

i strukture elektropolirovannoy poverkhnosti)

PERIODICAL: Doklady Akademii nauk SSSR, 1958, Vol. 120, Nr 1, pp.101-102

(USSR)

ABSTRACT: The problem of the mechanism of the electrolytic polishing

is directly connected with the problem of the structure of an electrically polished surface. The electromicroscopic investigation of such an electrically polished surface carried out by the authors (Ref 2) showed that the conception of

2 different stages of electro-polishing and especially the conception of the suppression of the structural blanching in the second stage are not at all beyond any doubts. The

present paper gives some results of these investigations. Electrically polished samples of polycrystalline copper were

Card 1/3 investigated. This electric polishing took place in an 5-M-

On the Problem of the Mechanism of the Electrolytic Polishing of Metals am of the Structure of the Polished Surface

-solution of phosphoric acid. Colloidal replica taken of the samples which had been electrically polished for were. different periods (10, 25 and 180 seconds) and then they were investigated by theans of the electron-microscope. According to these microphotographs in the first stage of electro--polishing (100 seconds) an active structural blanching takes place. The result of this blanching highly depends on the degree of the electrochemical inhomogeneity of the surface. A prolongation of the duration of electro-polishing leads to a further change of the character of structural blanching. A further increase of the duration of the electro-polishing to 180 seconds (i.e. the transition to that stage which is regarded the decisive stage of the process) does not bring about any important qualitative changes for the picture of structural blanching. The enclosed microphotographs do not tend to show in any way a suppression of the structural blanching during the total duration of the process. The mentioned data agree with those of other authors (Ref 4). There are 4 figures. and 17 references, 8 of which are Soviet.

Card 2/3

SOV/ 20-120-1-26/63

On the Problem of the Mechanism of the Electrolytic Polishing of Metals and of the Structure of the Polished Surface

ASSOCIATION: Khimicheskiy institut Kazanskogo filiala Akademii nauk SSSR

(Chemical Institute of the Kazan' Branch, AS USSR)

PRESENTED: January 2, 1958, by A. N. Frumkin, Member, Academy of Sciences,

USSR

SUBMITTED: November 18, 1957

1. Metals--Surface properties 2. Electrolytic polishing--Analysis

3. Surfaces--Structural analysis 4. Electron microscopes--Appli-

cations

Card 3/3

12 工作的证据

31548 S/081/61/000/022/011/076 B102/B108

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AUTHORS:

Valeyev, A. Sh., Gorbachuk, G. A.

Processes occurring in the range of the first rize of the TITLE:

polarization curve for anodic dissolution of steel in

electropolishing electrolyte

Referativnyy zhurnal. Khimiya, no. 22, 1961, 66, abstract PERIODICAL:

22B469 (Izv. Kazansk. fil. AN SSSR. Ser. khim. n., no. 5,

1959 61-69)

TEXT: The mechanism of anodic dissolution of a metal in an electropolishing electrolyte is studied. The measured polarization curves (PC) for carbon steel 50 and an electrolyte consisting of 70% H<sub>3</sub>PO<sub>4</sub> + 14% CrO<sub>3</sub> + 16% H<sub>2</sub>O, at 80°C are presented, as well as photomicrographs of the specimens treated under conditions corresponding to different points of the PC. The PC showed two sections of current rise and a range of a limiting current. The cause of the dull etching of the specimens in the range of the first current rise is investigated. It is assumed that under these conditions the passivating film is incomplete which leads to Card 1/2

Processes occurring in the range ...

31548 \$/081/61/000/022/011/076 B102/B108

microcell operation, causing separation of a loose Fe layer in the cathode region. At i <0.7 a/dm², the Fe layer is formed under a thin metal film (RZhKhim, 1956, No. 23, 75605), at i>0.7 a/dm² the number of active region increases, and the highly disperse Fe is deposited on the whole surface of the specimen. It is pointed out that redox reactions in the electrode-new regions, e.g. interaction between Fe²+ and Cr₂0² have to be taken into account. In the authors' opinion the proposed etching mechanism verifies their own data on the dependences of the effective metal yield with respect to current and losses in weight of the specimens owing to anodic dissolution and self-dissolution on the composition of the electrolyte, on note: Complete translation. Abstracter's

Card 2/2

\$/020/60/133/04/25/031 B004/B056

AUTHORS: Yozdvizhenskiy, G. S., Gorbachuk, G. A., Dezider'yev, G.

TITLE: The Mechanism of the Electrolytic Polishing of Metals in

the Light of Electron-microscopic Studies of the Surface

During the Various Stages of Treatment

PERIODICAL: Doklady Akademii nauk SSSR, 1960, Vol. 133, No. 4,

pp. 869 - 871

TEXT: In an earlier paper (Ref. 4), the authors proved that in the anodic dissolution of metals a structural etching of the surface always occurs. In the present paper, the connection between anodic dissolution and metal structure was investigated. Cold-rolled copper sheets of the type M1 (M1) and cold-rolled brass sheets of the type J159 (L59) were used as samples. Electrolytic polishing was carried out in 11.35 M orthophosphoric acid at a current density of 14.5 ma/cm<sup>2</sup> for copper and 19 ma/cm<sup>2</sup> for brass. The samples were first etched until a distinct microstructure became visible (copper with ammoniumpersulfate, brass with an aqueous solution of ammonia and hydrogen peroxide). After electrolytic polishing

Card 1/2

The Mechanism of the Electrolytic Polishing S/020/60/133/04/25/031 of Metals in the Light of Electron-microscopic B004/B056 Studies of the Surface During the Various Stages of Treatment

of different periods of time, the surfaces were examined metallegraphically (200-fold magnification) and under an electron microscope (23,000-fold magnification) (Figs. 1, 2). The same experiments were carried out with copper and brass sheets annealed in vacuo (Figs. 3, 4). It follows from Figs. 1, 2 that in electrolytic polishing the surface structure orientated by rolling becomes clearly visible. By annealing, this structural orientation is again lost. A so-called suppression of structural etching does, however, not occur. Only the extent of anodic dissolution changes in accordance with the change in electrochemical inhomogeneity. Here, diffusion processes play an important part in that the differences in the metal relief caused by etching are gradually equalized. There are 4 figures and 4 references: 3 Soviet, 1 American, 1 Canadian, and 1 German.

ASSOCIATION: Khimicheskiy institut Kazanskogo filiala Akademii nauk SSSR (Chemical Institute of the Kazan' Branch of the Academy of

Sciences USSR)

PRESENTED: January 28, 1960 by A. N. Frumkin, Academician

SUBMITTED: January 28, 1960

Card 2/2

S/137/62/000/006/138/163 A057/A101

AUTHORS:

Vozdvizhenskiy, G. S., Gorbachuk, G. A., Dezider'yev, G. P.

TITLE:

Electron-microscopic investigation of the process of anodic decrys-

tallization of a metal

PERIODICAL: Referativnyy zhurnal, Metallurgiya, no. 6, 1962, 90 - 91, abstract 61574 (V sb. "Rost kristallov, T. Z.", Moscow, AN SSSR, 1961, 192 -

199. Discus., 214 - 218)

The process of anodic dissolving of rolled and annealed Cu was studied in 5% H3r04 solution. The Cu-samples were etched in 10% ammonium persulfate solution, then inserted into the electrolytic bath, separated and studied metallographically and electron-microscopically. The initial surface of such samples is electrochemically non-uniform, thus after chemical etching a crystallographic structure is revealed. The anodic dissolving of the sample occurs under participation of an acceptor, which is included in the composition of the electrolyte. Microphotographs showed that the outer effect of etching in ammonium persulfate upon the rolled material corresponds to the orientation of

Card 1/2

Electron-microscopic investigation...

S/137/62/000/006/138/163 A057/A101

crystals in the direction of rolling; this is even more pronounced in electro-chemical treatment. The electron-microscopic pictures reveal numerous oriented shifts inside the crystals. The annealed material shows the same picture of chemical and electrochemical dissolving as the rolled material. Since the annealed material contains considerably larger crystallites than the rolled material, the effect of chemical and electrochemical dissolving is manifested at smaller enlargements. Annealing effects desorientation of crystallites, thus the treated elements of surface do not show an orientation either on optical, or on electron-microscopical photographs. As a result of anodic dissolving of metals, the picture of the formation and growth of crystallites is clearly developed, which is a proof of the community of the electrodecrystallization mechanism at anodic dissolving of metals. There are 6 references.

Ye. Layner

[Abstracter's note: Complete translation]

Card 2/2

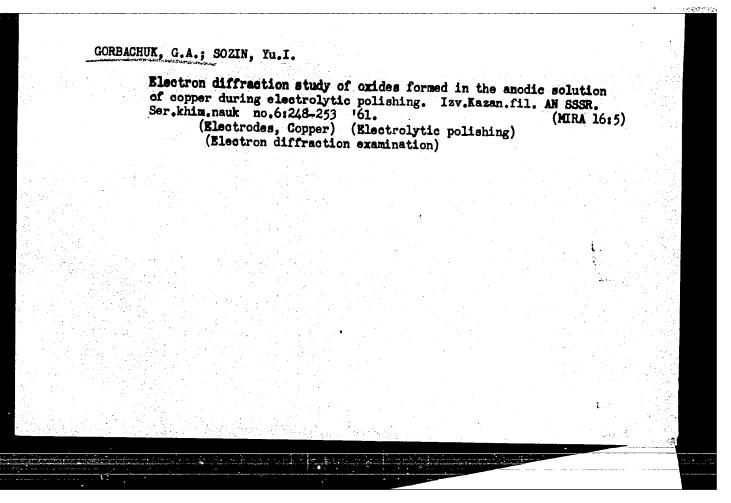
Formation of an oxide layer in the course of the electrolytic polishing of copper. Izv.Kazan.fil. AN SSSR. Ser.khim.nauk					
no.6:155-162 '61. (Copper—Finishing)	(Electrolytic polishing)	(MIRA 16:5) (Metallic oxides)			

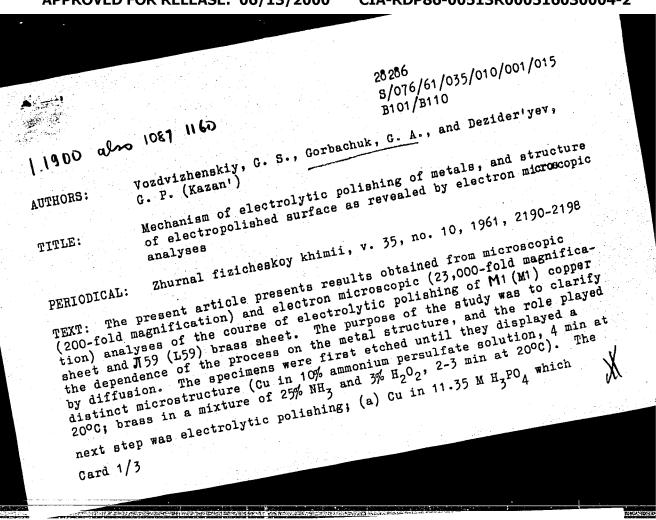
VOZDVIZHENSKIY, G.S.; GORBACHUK, G.A.; DEZIDER'YEV, G.P.

Mechanism of the electrolytic polishing of metals and the structure of the polished surface according to electron microscopy data.

Izv.Kasan.fil. AN SSSR. Ser.khim.nauk no.6:129-143 '61. (MIRA 16:5) (Electrolytic polishing) (Electron microscopy)

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Mechanism of electrolytic ...

contained 4.8 g/liter of Cu, at 14.5 ma/cm2; (b) brass in 11.35 M H<sub>3</sub>PO<sub>4</sub> which contained 2.9 g/liter of Cu, at 19 ma/cm2. These optimum concentrations led to reproducible results. The reflection factor of the polished sheet was determined by putting the reflection coefficient of a silver mirror = 100%. The reflection factor was found to attain a high value, as soon as the anode potential was stable. Replicas of specimens polished between 3 and 100 min were examined in an 3M-3 (EM-3) electron microscope. A second set of experiments was performed with annealed specimens made of the same metals. Cu was annealed for 1.5 hr at 700°C, and brass for 2 hr at 600°C. The specimens were polished in an electrolyte with a concentration like that of the 1st set. The current density, however, was 21 ma/cm<sup>2</sup> for Cu, and 16 ma/cm<sup>2</sup> for brass. The annealed specimens displayed an inhomogeneous surface with disoriented crystallites. In them, reflection factor and anode potential had lower values than in non-annealed metal. The processes observed are explained by deep etching figures being formed at first (after 3-5 min). Penetration into their depths is however, soon inhibited, and the respective are passivated. The crystals are then dissolved breadthwise. Electron microscopic analyses showed that the original texture of rolled (oriented) and annealed (disoriented) metal Card 2/3

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remained unchanged in this connection. The surface is smoothed by a decrystallization process (decomposition of crystallites). The anode passivation is a consequence, not the cause, of this process by which the electrochemical inhomogeneity is balanced. A paper by S. I. Krichmar (Dokl. AN SSSR, 122, 424, 1958) is mentioned. There are 4 figures, 4 tables, and 20 references: 15 Soviet and 5 non-Soviet. The two most important references to English-language publications read as follows: G. Lucien, J. Andre, J. Phys. Chem., 57, 701, 1953; J. Edwards, J. Electrochem. Soc., 100, 189, 1953.

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